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SYNTHESES OF 1,3-DISUBSTITUTED IMIDAZO[1,5-a]PYRIDINES

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<u>Abstract</u>: Facile syntheses of 1,3-disubstituted imidazo[1,5-a]pyridines have been effected by treatment of ketimines derived from di-2-pyridyl ketone with LDA and benzophenone, reactions of benzyl amine with di-2-pyridyl ketone in boron trifluoride etherate, and reaction of di-2-pyridyl methanamine imines with molecular sieves in benzene.

We wish to report several facile synthetic routes to 1,3-disubstituted imidazo[1,5-a] pyridines $\underline{1}$ which were uncovered during a study of the regiochemistry of electrophilic attack on anions derived from N-alkyl diarylketimines. 1,2

a, R = 2-pyridy1, R¹ =
$$CH_2CH_3$$

b, R = 2-pyridy1, R¹ = $CH(CH_3)_2$
c, R = 2-pyridy1, R¹ = C_6H_5
d, R = R¹ = C_6H_5
e, R = 2-pyridy1, R¹ = CH_3
f, R = R¹ = 2-pyridy1

Treatment of imine $\underline{2}a$ with LDA at $-78^{\circ}C$ in THF, followed by a quench with 1N HC1, led to the amine $\underline{3}$ (74%). On the other hand, when $\underline{2}a$ or $\underline{2}b$ were treated with LDA at $-78^{\circ}C$ in THF, followed by the addition of one equivalent of benzophenone, imidazo[1,5-a]pyridines $\underline{1}a$ (88%) $\underline{3}$, $\underline{4}$ and $\underline{1}b$ (66%) were isolated, respectively. In the reaction leading to $\underline{1}a$, a quantitative yield of benzhydrol was also obtained. The anions derived from the imines $\underline{2}a$ or $\underline{2}b$ underwent cyclizations and subsequent oxidations by benzophenone to yield $\underline{1}a$ or $\underline{1}b$.

NH2

N a, Ar = 2-pyridyl

b, Ar =
$$C_6H_5$$

Most prior syntheses of imidazo [1,5-a]pyridines involve the formation of the imidazo ring by dehydrative cyclizations of amides derived from 2-aminomethylpyridines 4,5 and related compounds. 6,7

In an attempt to synthesize $\underline{2}c$, by heating ketone $\underline{4}a$ with benzylamine in the presence of boron trifluoride etherate, it was found that the major product of this reaction was $\underline{1}c$ (46%). A small amount of the expected imine $\underline{2}c$ (10%) was also isolated. In a similar manner, ketone $\underline{4}b$ reacted with benzylamine to give $\underline{1}d$ (39%). The imines $\underline{2}c$ and $\underline{5}$ which are initially formed undergo rearrangement to the corresponding isomeric imines $\underline{6}a$ and $\underline{6}b$, respectively, which then undergo cyclizations followed by air oxidation to yield the imidazo[1,5-a]pyridines.

To test this hypothesis, the imines $\underline{6}c$ and $\underline{6}d$ were prepared from $\underline{3}$ by reactions with the appropriate aldehydes. These imines underwent cyclizations in the presence of molecular sieves in benzene solutions at room temperature to yield $\underline{1}e$ (39%) and $\underline{1}f$ (47%), $\underline{8}$ respectively.

CHR a, Ar = 2-pyridyl, R =
$$C_6H_5$$
b, Ar = R = C_6H_5
c, Ar = 2-pyridyl, R = CH_3
d, Ar = R = 2-pyridyl

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References and Notes

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- 2. Kauffmann, T.; Koppelmann, E.; Berg, H. Angew. Chem. Internat. Edit. 1970, 9, 163.
- 3. $\underline{1}$ a; mp 77-78°C; PMR (CDC1₃) δ 1.48 (3H, t), 2.07 (2H, q), 6.73 (1H, m), 6.87 (1H, m), 7.04 (1H, m), 7.65-7.80 (2H, m), 8.12 (1H, d), 8.60 (2H, m). Anal. Calcd for $C_{1\Delta}H_{13}N_3$: %C, 75.31; %H, 5.87; %N, 18.82: Found: %C, 75.06; %H, 5.97; %N, 18.81.
- 4. An authentic sample of <u>la</u> was prepared by the procedure of Bower and Ramage [<u>J. Chem. Soc.</u> 1955, 2834] from di-(2-pyridy1)methanamine(<u>3</u>) and propionyl chloride.
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- 6. Katritzky, A. P.; Rees, C. W. Comprehensive Heterocyclic Chem. 1984, 5, 634.
- 7. Edgar, M. T.; Pettit, G. R.; Krupa, T. S. <u>J. Org. Chem.</u> 1979, <u>44</u>, 396. This report describes the isolation of an imidazo[1,5-a]pyridine (8%) in the attempted synthesis of an N-benzylidene 2-pyridyl amino acid.
- 8. <u>1f</u>; mp 147-147.5°C; PMR (CDCl₃)δ 6.83 (1H, m), 7.03-7.15 (2H, m), 7.22 (1H, m), 7.78 (2H, m), 8.30 (1H, m), 8.49 (1H, m), 8.64 (2H, m), 8.78, (1H, m), 10.03, (1H, m). Anal. Calcd. for C₁₇H₁₂N₄: %C, 74.98; %H, 4.44; %N, 20.57: Found: %C, 74.87; %H, 4.71; %N, 20.32.

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